| C3 | $-0.0451(4)$ | $0.0625(2)$ | $0.7684(4)$ | $0.041(1)$ |
| :--- | ---: | ---: | :--- | :--- |
| C4 | $-0.0102(4)$ | $0.1326(2)$ | $0.7594(4)$ | $0.046(1)$ |
| C5 | $0.1505(3)$ | $0.2192(2)$ | $0.8507(5)$ | $0.049(1)$ |
| C6 | $0.2263(4)$ | $0.2457(2)$ | $0.9897(6)$ | $0.062(1)$ |
| C7 | $0.3162(4)$ | $0.1954(2)$ | $1.0602(5)$ | $0.062(2)$ |
| C8 | $0.2403(3)$ | $0.1336(2)$ | $1.0990(4)$ | $0.048(1)$ |
| C9 | $0.1703(3)$ | $0.1059(2)$ | $0.9554(4)$ | $0.038(1)$ |
| C10 | $0.0747(3)$ | $0.1575(2)$ | $0.8909(4)$ | $0.040(1)$ |
| C11 | $0.1827(3)$ | $-0.0089(2)$ | $1.0204(4)$ | $0.045(1)$ |
| C12 | $0.1204(3)$ | $-0.0736(2)$ | $1.0540(4)$ | $0.043(1)$ |
| C13 | $0.1805(4)$ | $-0.1281(2)$ | $0.9907(4)$ | $0.054(1)$ |
| C14 | $0.1260(5)$ | $-0.1890(2)$ | $1.0121(5)$ | $0.068(2)$ |
| C15 | $0.0131(5)$ | $-0.1963(2)$ | $1.1009(6)$ | $0.070(2)$ |
| C16 | $-0.0436(5)$ | $-0.1433(2)$ | $1.1711(5)$ | $0.068(2)$ |
| C17 | $0.0093(4)$ | $-0.0822(2)$ | $1.1475(5)$ | $0.055(1)$ |
| C21 | $-0.1413(3)$ | $0.0614(1)$ | $1.0438(4)$ | $0.038(1)$ |
| C22 | $-0.1116(4)$ | $0.0849(2)$ | $1.1913(4)$ | $0.047(1)$ |
| C23 | $-0.2121(5)$ | $0.1085(2)$ | $1.2875(5)$ | $0.063(2)$ |
| C24 | $-0.3427(4)$ | $0.1079(2)$ | $1.2391(5)$ | $0.069(2)$ |
| C25 | $-0.3738(4)$ | $0.0835(2)$ | $1.0951(6)$ | $0.066(2)$ |
| C26 | $-0.2747(3)$ | $0.0603(2)$ | $0.9979(5)$ | $0.050(1)$ |

Table 2. Selected geometric parameters ( $\left(\AA,{ }^{\circ}\right.$ )

| $\mathrm{O} 4-\mathrm{C} 4$ | $1.210(5)$ | $\mathrm{N} 1-\mathrm{C} 2$ | $1.475(4)$ |
| :--- | ---: | :--- | ---: |
| $\mathrm{O} 11-\mathrm{C} 11$ | $1.235(4)$ | $\mathrm{N} 1-\mathrm{C} 9$ | $1.484(4)$ |
| $\mathrm{O} 4-\mathrm{C} 4-\mathrm{C} 3$ | $122.1(3)$ | $\mathrm{C} 13-\mathrm{C} 12-\mathrm{C} 17$ | $118.2(4)$ |
| $\mathrm{C} 3-\mathrm{C} 4-\mathrm{C} 10$ | $115.1(3)$ | $\mathrm{C} 2-\mathrm{C} 21-\mathrm{C} 26$ | $118.9(3)$ |
| $\mathrm{O} 4-\mathrm{C} 4-\mathrm{C} 10$ | $122.8(4)$ | $\mathrm{C} 2-\mathrm{C} 21-\mathrm{C} 22$ | $123.1(3)$ |
| $\mathrm{C} 11-\mathrm{Cl}-\mathrm{C} 17$ | $124.0(4)$ | $\mathrm{C} 22-\mathrm{C} 21-\mathrm{C} 26$ | $118.0(3)$ |
| $\mathrm{C} 11-\mathrm{Cl} 2-\mathrm{C} 13$ | $117.7(3)$ |  |  |
| $\mathrm{C} 2-\mathrm{N} 1-\mathrm{C} 11-\mathrm{C} 12$ | $-22.8(5)$ | $\mathrm{O} 4-\mathrm{C} 4-\mathrm{C} 10-\mathrm{C} 9$ | $-148.9(4)$ |
| $\mathrm{C} 2-\mathrm{N} 1-\mathrm{C} 9-\mathrm{C} 10$ | $11.0(4)$ | $\mathrm{C} 6-\mathrm{C} 5-\mathrm{C} 10-\mathrm{C} 9$ | $-56.7(4)$ |
| $\mathrm{C} 11-\mathrm{N} 1-\mathrm{C} 9-\mathrm{C} 8$ | $-69.1(4)$ | $\mathrm{C} 10-\mathrm{C} 5-\mathrm{C}-\mathrm{C} 7$ | $54.0(5)$ |
| $\mathrm{C} 2-\mathrm{N} 1-\mathrm{C} 9-\mathrm{C} 8$ | $133.7(3)$ | $\mathrm{C} 5-\mathrm{C} 6-\mathrm{C}-\mathrm{C} 8$ | $-53.5(5)$ |
| $\mathrm{C} 9-\mathrm{N} 1-\mathrm{C} 2-\mathrm{C} 3$ | $41.4(4)$ | $\mathrm{C} 6-\mathrm{C} 7-\mathrm{C} 8-\mathrm{C} 9$ | $56.6(4)$ |
| $\mathrm{N} 1-\mathrm{C} 2-\mathrm{C} 21-\mathrm{C} 22$ | $-6.7(5)$ | $\mathrm{C} 7-\mathrm{C} 8-\mathrm{C} 9-\mathrm{C} 10$ | $-58.2(4)$ |
| $\mathrm{N} 1-\mathrm{C} 2-\mathrm{C} 3-\mathrm{C} 4$ | $-59.2(4)$ | $\mathrm{N} 1-\mathrm{C} 9-\mathrm{C} 10-\mathrm{C} 4$ | $-49.4(4)$ |
| $\mathrm{C} 3-\mathrm{C} 2-\mathrm{C} 21-\mathrm{C} 22$ | $-131.7(4)$ | $\mathrm{C} 8-\mathrm{C} 9-\mathrm{C} 10-\mathrm{C} 5$ | $58.2(4)$ |
| $\mathrm{C} 2-\mathrm{C} 3-\mathrm{C} 4-\mathrm{C} 10$ | $21.9(5)$ | $\mathrm{N} 1-\mathrm{C} 11-\mathrm{C} 12-\mathrm{C} 13$ | $139.8(3)$ |
| $\mathrm{C} 3-\mathrm{C} 4-\mathrm{C} 10-\mathrm{C} 5$ | $158.3(3)$ | $\mathrm{O} 11-\mathrm{C} 11-\mathrm{C} 12-\mathrm{C} 13$ | $-39.8(5)$ |
| $\mathrm{C} 3-\mathrm{C} 4-\mathrm{C} 10-\mathrm{C} 9$ | $31.8(4)$ |  |  |

The structure was solved by direct methods using SHELXS86 (Sheldrick, 1985). All non-H atoms were refined with anisotropic displacement parameters. All H atoms were obtained from difference Fourier maps and were included in the structure-factor calculations; they were given displacement parameters equal to $1.1 U_{\mathrm{eq}}$ of their respective carrier atom, but their parameters were not refined (Sheldrick, 1976). The geometrical calculations were performed using PARST (Nardelli, 1983).

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# $o$-Phenylenediammonium Bis(hydrogensulfide) 

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## Abstract

The title compound, $\mathrm{C}_{6} \mathrm{H}_{10} \mathrm{~N}_{2}^{2+} .2 \mathrm{HS}^{-}$, forms crystals with $\mathrm{C} 2 / \mathrm{c}$ symmetry. The anionic $\mathrm{HS}^{-}$groups are located near the $\mathrm{NH}_{3}^{+}$substituents and are oriented practically perpendicular to the benzene ring.

## Comment

The molecule in the unit cell of the title compound, (I), is situated on the twofold axis, which passes through the midpoints of the $\mathrm{C}(4 A)-\mathrm{C}(4 B)$ and $\mathrm{C}(1 A)-\mathrm{C}(1 B)$ bonds.

(I)

The presence of H atoms at every N atom and the $\mathrm{C}(4 A)-\mathrm{N}(1 A)$ bond length of $1.457(2) \AA$ indicate protonation of both amino groups. The anionic HS ${ }^{-}$groups are oriented practically perpendicular to the benzene ring [ $\left.104(1)^{\circ}\right]$. The short interatomic distances $\mathrm{N}(1 A) \cdots \mathrm{S}(A)[3.11(1) \AA]$ and $\mathrm{H} 1(\mathrm{~N} 1 A) \cdots \mathrm{S}(A)$ [ $2.13(1) \AA$ ] confirm the existence of a strong electrostatic interaction between the dication and the anions.

There are no intermolecular distances shorter than the sum of the van der Waals radii. Fig. 1 shows a perspective view of (I), with the atom-numbering scheme.


Fig. 1. A view of the title molecule with the atom-numbering scheme. Atoms are represented as $50 \%$ probability ellipsoids.

## Experimental

Dark red crystals suitable for X-ray study were grown from a 2-propanol solution by slow evaporation of the solvent.

## Crystal data

$\mathrm{C}_{6} \mathrm{H}_{10} \mathrm{~N}_{2}^{2+} .2 \mathrm{HS}^{-}$
$M_{r}=176.31$
Monoclinic
C2/c
$a=7.341$ (1) $\AA$
$b=14.518$ (3) $\AA$
$c=8.010(2) \AA$
$\beta=94.01$ (3) ${ }^{\circ}$
$V=851.6(9) \AA^{3}$
$Z=4$
$D_{x}=1.375 \mathrm{Mg} \mathrm{m}^{-3}$

## Data collection

Siemens P3/PC diffracto eter
$2 \theta-\theta$ scans
Absorption correction: none
1917 measured reflections
1804 independent reflections 1119 observed reflections

$$
[F>6 \sigma(F)]
$$

## Refinement

Refinement on $F$
$R=0.037$
$w R=0.043$
$S=1.73$
1119 reflections
64 parameters
H atoms riding with fixed isotropic $U$

Mo $K \alpha$ radiation
$\lambda=0.71073 \AA$
Cell parameters from 9 reflections
$\theta=12-13^{\circ}$
$\mu=0.554 \mathrm{~mm}^{-1}$
$T=293 \mathrm{~K}$
Block
$0.4 \times 0.2 \times 0.2 \mathrm{~mm}$
Dark red
$R_{\text {int }}=0.0259$
$\theta_{\text {max }}=25^{\circ}$
$h=0 \rightarrow 11$
$k=0 \rightarrow 23$
$l=-12 \rightarrow 12$
2 standard reflections monitored every 98 reflections intensity decay: 7\%
$w=1 /\left[\sigma^{2}\left(F_{o}\right)+0.0003 F_{o}^{2}\right]$
$(\Delta / \sigma)_{\text {max }}=0.197$
$\Delta \rho_{\text {max }}=0.44 \mathrm{e}_{\AA^{-3}}$
$\Delta \rho_{\min }=-0.47 \mathrm{e}^{\AA^{-3}}$
Extinction correction: none
Atomic scattering factors from International Tables for X-ray Crystallography (1974, Vol. IV)

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters $\left(\AA^{2}\right)$

$$
U_{\mathrm{eq}}=(1 / 3) \Sigma_{i} \Sigma_{j} U_{i j} a_{i}^{*} a_{j}^{*} \mathbf{a}_{i} \mathbf{a}_{j}
$$

|  | $x$ | $y$ | $z$ | $U_{\text {eq }}$ |
| :--- | :---: | :---: | :---: | :---: |
|  | $\boldsymbol{y}$ |  |  |  |
| $\mathbf{S}$ | $0.0232(1)$ | $0.1767(1)$ | $-0.0099(1)$ | $0.037(1)$ |
| $\mathrm{N}(1)$ | $-0.1880(2)$ | $0.3323(1)$ | $-0.2069(2)$ | $0.035(1)$ |
| $\mathrm{C}(1)$ | $-0.0885(3)$ | $0.5828(1)$ | $-0.2253(2)$ | $0.053(1)$ |
| $\mathrm{C}(2)$ | $-0.1787(2)$ | $0.5008(1)$ | $-0.2017(2)$ | $0.042(1)$ |
| $\mathrm{C}(4)$ | $-0.0889(2)$ | $0.4181(1)$ | $-0.2264(2)$ | $0.029(1)$ |

Table 2. Selected geometric parameters $\left(\AA^{\circ},^{\circ}\right)$

| $\mathrm{C}(4)-\mathrm{C}\left(4^{i}\right)$ | $1.384(2)$ | $\mathrm{C}(4)-\mathrm{C}(2)$ | $1.391(2)$ |
| :--- | :---: | :--- | :--- |
| $\mathrm{C}(1)-\mathrm{C}\left(1^{1}\right)$ | $1.384(4)$ | $\mathrm{C}(2)-\mathrm{C}(1)$ | $1.381(2)$ |
| $\mathrm{N}(1)-\mathrm{C}(4)$ | $1.457(2)$ |  |  |
| $\mathrm{N}(1)-\mathrm{C}(4)-\mathrm{C}(2)$ | $118.6(1)$ | $\mathrm{C}(4)-\mathrm{C}(2)-\mathrm{C}(1)$ | $119.2(1)$ |
| $\mathrm{C}(2)-\mathrm{C}(4)-\mathrm{C}\left(4^{\prime}\right)$ | $120.3(1)$ | $\mathrm{C}(2)-\mathrm{C}(1)-\mathrm{C}\left(1^{\prime}\right)$ | $120.5(1)$ |
| $\mathrm{N}(1)-\mathrm{C}(4)-\mathrm{C}\left(4^{i}\right)$ | $121.1(1)$ |  |  |
| Symmetry codes: (i) $-x, y,-\frac{1}{2}-z$ |  |  |  |

The structure was solved by direct methods using SHELXTLPlus (Sheldrick, 1991). After non-H atoms were refined anisotropically, positions of all H atoms were located from a $\Delta F$ map and included in the refinement with fixed isotropic displacement parameters. Ten strong reflections with ( $F_{o^{-}}$ $\left.F_{c}\right) / \sigma>4.0$ were excluded from the last refinement cycles.

Lists of structure factors, anisotropic displacement parameters, Hatom coordinates and complete geometry have been deposited with the IUCr (Reference: VS1009). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square. Chester CH1 2HU, England.

## References

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## Muscarinic Antagonist: 8,8-Dimethyl-3' $\mathbf{3}^{\prime} \mathbf{3}^{\prime}$ -diphenylspiro(8-azoniabicyclo[3.2.1]octane$3,2^{\prime}$ - $\mathbf{1}^{\prime}, 3^{\prime}$-dioxolane) $-4^{\prime}$-one Iodide

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## Abstract

The molecular structure of the title compound, $\mathrm{C}_{23} \mathrm{H}_{26} \mathrm{NO}_{3}^{+} . \mathrm{I}^{-}$, BVT44Me, has been compared to that of the related compound 8,8-dimethyl-3,3-diphenyl-

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[^0]:    Lists of structure factors, anisotropic displacement parameters, H-atom coordinates, complete geometry, least-squares-planes data and torsion angles have been deposited with the IUCr (Reference: VJ1020). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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